Improved Corrosion and Abrasion Resistance of Organic-Inorganic Composite Coated Electro-galvanized Steels for Digital TV Panels

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Summary

Recently, household electronic industries require environmentally-friendly and highly functional steels to enhance quality of human life. Especially, customers require both excellent corrosion and abrasion resistant anti-fingerprint steels for digital TV panels. Thus POSCO have developed new functional electro-galvanized steels, which are coated double layer with organic-inorganic composite on the zinc surface of steel, for usage as the bottom chassis panel of TVs. The inorganic solution for the bottom layer consists of inorganic phosphate, magnesium, and zirconium compound with a small amount of epoxy binder, and affords both improvedadhesion properties by chemical conversion reaction and corrosion resistance due to a self-healing effect. The composite solution for the top layer was prepared by fine dispersion of organic-inorganic ingredients that consist of urethane modified polyacrylate polymer, hardener, silica sol and a titanium complex inhibitor in aqueous media. Both composite solutions were coated on the steel surface by using a roll coater and then cured through induction furnace in the electro-galvanizing line. New antifingerprint steels were evaluated for quality performances such as salt spray test for corrosion resistance, tribological test for abrasion resistance, and conductivity test for surface electric conductance regarding to both types of polymer resin and coating weight of composite solution. New composite coated anti-fingerprint steels afford both better corrosion resistance and abrasion properties compared to conventional anti-fingerprint steels that mainly consist of acrylate polymer. Detailed discussions of both composite solution and experimental results suggest that urethane modification of acrylate polymer of composite solution plays a key role of enhanced quality performances.

1. Introduction

To date, zinc and zinc-alloy coated steel sheets have been treated using both chromate chemical conversion and/or chromate-free coating on the zinc surface to enhance corrosion resistance and additional functional properties. Typically, chromate treatment is effective for rust-prevention on a zinc layer which is widely used because of both performance and economical aspects. The most important reason for the excellent performance of the chromate coating seems to be the polymeric network in the layer through the bridge between Cr(VI) and the oxo-/ hydroxo- anion.[1] Steel makers have used chromate as a protective coating substance of zinc coated steels for a long time. But several years ago, the EPA classified the chemicals as a human carcinogen when inhaled, and heavily regulates their use and disposal. Especially, EU directives have a profound impact on the protective coating for metals. From July 2006, EU directives have targeted the electrical and electronic equipment industries with further restrictions.[2-3] For the last decade, POSCO have made an effort to develop chromate-free galvanizing steels according to the previously described situation.[4] These chromate-free products are manufacturing and widely using as a materials in household, construction, and automotive steels. In particular,

replacing the chromate conversion coating on the anti-fingerprint electro-galvanized steels has been developed by introducing organic-inorganic composite technology,[5-7] as shown in Scheme 1. However, customers require both excellent corrosion and abrasion resistant anti-fingerprint steels for use as deep drawing parts of digital TV panels. Therefore in this paper, we describe development and quality performances of environmentally-benign, double layered and composite coated electro-galvanized steels as a substitute for the conventional anti-fingerprint analogs.



2. Experimental Details

In this section, we describe the preparation methods of the organic-inorganic composite solution, surface coated steels, and quality performances.

2.1 Preparation of Composite Solution. The composite coating solutions were prepared as an aqueous solution that comprised of both inorganic solutions for bottom layer and organic-inorganic solution for top layer. The former solution consists of an inorganic phosphate, silicate, and manganese compound. The latter composite solution was prepared from the mechanical dispersion of acryl modified polyurethane binder, melamine hardener, micronized silica, titanate compound as an anticorrosive agent, and fluorine substituted polyethylene wax as a lubricant, respectively. Several miscellaneous additives were added into the resin solution; wetting and stabilization agent of pigments, defoamer and so on. The solid content was measured by weight after drying at 150 °C for 30 min. and viscosity of the composite solution was measured by Brookfield method, respectively.

2.2 Production of Composite Coated Steels. The organic-inorganic composite solutions for mass production were prepared from water-borne solutions that had 5 ± 1 cps for the bottom layer and 8 ± 1 cps for the top layer in viscosity as well as 8 ± 1 (bottom) and 16 ± 1 (top) % in solid content at 25 °C. The production was carried out in the electro-galvaning line (*denotes* EGL) that the processes consisted of pre-treatment, galvanizing, roll coating, induction curing and cooling section, according in order. The composite solution was coated by three-roll reverse operation on the strip surface.[8] The curing temperature was set at 120 ± 5 °C for the bottom layer and 160 ± 5 °C for the top layer by induction oven. Both coated layers were dried in an induction oven and then cooled to room temperature *via* air cooling and water quenching section. The weight of the coated layer was measure with both a portable gauge of infrared wavelength measurement instrument and wet weight method.

2.3 Performance Evaluation. The quality performances for the surface of anti-fingerprint steels were evaluated by the following procedure. The corrosion resistance was evaluated by salt spray test (SST) with edge sealing by scotch tape for both plain and Ericksen (7mm) drawn parts.[9] Anti-fingerprint property was measured by color difference before and after

by applying artificial tears solution (JIS-K2246) for 5 seconds. Solvent resistance was evaluated from color difference before and after by applying organic solvent such as isopropyl alcohol, methylethylketone, methylene chloride and toluene by rubbing 10 times at load force of 0.25 kgf/mm². The electric conductivity was evaluated by surface resistance which was measured by four-pin probe method (model, Loresta GP). Thermal resistance was measured by color differences after 1 hour of heating in the oven at a temperature of 250 °C. The friction coefficient was measured from draw bead tester at measurement conditions of loading weight 600 kgf, speed 150 mm/min and length 100 mm without a lubricant. Abrasion property was tested and rated into grades after rubbing 10 times at load force of 0.25 kgf/mm² without a lubricant.

3. Results and Discussion

In general, anti-fingerprint steels require basic properties such as good corrosion resistance, surface electric conductivity, and abrasion resistance for panel usage of home electronic appliances.[4] Among these characteristics in the organic polymer-based resin coated steels, there are good correlations with coating weights on the zinc surface shown in Figure 1. As the coating weight increases, electrical conductivity decreases due to the insulating property of the coating layer, however anti-corrosive property increases because of the increased barrier effect of the thicker coating layer. As the coating layer, however anti-corrosive property increases, electrical conductivity decreases due to insulation property of the coating layer, however anti-corrosive property increases because of increased barrier effect of the thicker coating layer. As the coating layer, however anti-corrosive property increases because of increased barrier effect of the thicker coating layer. As though higher coating weight gives higher abrasion resistance that affords better press formability, but for the household appliances a proper coating weights (or thickness) of the Cr-free coating layer are the most important to get excellent electrical conductivity to enhance electromagnetic interference shielding inside instrument. Generally these characteristics tend to change drastically at about 1000mg/m² region by coating weights.



Figure 1. Typical correlation between the coating weights vs. electrical conductivity, corrosion resistance, and abrasion property on the organic polymer-based resin coated steels.

3.1 Composite Coated Steel Sheets. New Cr-free composite technology achieves good results by creating a thin organic-inorganic hybrid layer on the zinc coated surface that replaces the chromate historically used in commercial steels. The concept of the Cr-free double coating layer was designed by mimic polymeric network structure of chromate

layer.[1] Actually, chromate provides excellent anticorrosion properties from both the protective and self-healing effects that arise from the inorganic polymeric structure and high valent chromium ion. In our research, to afford environmentally-friendly steels, the concept of Cr-free composite coating layer was designed by double layered coating. Double layer afford both adhesion property between metal surface and top coating by inorganic bottom layer and abrasion property by organic-inorganic top layer. The former solution consists of inorganic phosphate, magnesium, and zirconium compound with a small amount of epoxy binder, and affords both excellent adhesion properties by chemical conversion reaction and corrosion resistance owing to the self-healing effect. The coating weights were set to 250 to 350 mg/m^2 to provide both excellent surface quality and productivity, as shown in Figure 3. The latter composite solution was prepared from the mechanical dispersion of acryl modified polyurethane binder, melamine hardener, micronized silica, titanate compound as an anticorrosive agent and fluorine substituted polyethylene wax as a lubricant, respectively. Several miscellaneous additives were added into the resin solution; wetting and stabilization agent of pigments, defoamer as well as others. The resin of the top layer was modified by acrylate modified urethane copolymer (molecular weight (MW) ~6x10³ and ~10⁴ (composition, 1:1), $T_q = 0$ and 10 °C, respectively) which provides both enhanced chemical and thermal resistance due to strong binding with the bottom layer compared to conventional ethylene acrylate polymer (MW ~10⁴, $T_{a} = -10$ °C). While the lower MW polymer provides both increased solution flow property and wetting on the metal surface, the higher molecular weight of - polymer affords better hardness due to toughness of the layer. The conceptual description of the coating layer before and after curing was shown in Figure 3. Also, the coating weight was adjusted by 600 to 800mg/m2 to satisfy both electrical conductivity and abrasion resistance, shown in Figure 2. Therefore new composite coated steels are developed double layer coated steels with 1000±150 mg/m² coating weights by a proper combination between the bottom and top layer composite coating solutions. As shown in Figure 2, both left and below the target coating weight describes alarming region for poor corrosion resistance due to short coating weight. While both right and upper of target coating weight describes alarming region for poor electrical conductivity owing to high coating weight.



Figure 2. Coating weight diagram based on correlation between bottom and top layer.



Figure 3. Conceptual description of top coating layer between conventional polyacrylate, (a) & (b), and new polyurethane, (c) & (d), before and after curing. Wet films, (a) & (c); dried films, (b) & (d).

3.2 Quality Performance. As the thickness of the coated layer goes to thin, the quality of the layer such as corrosion resistance and press formability is increasingly important. To improve both qualities in this development, small (MW ~6000) and high (MW ~10000) molecular weight polyurethanes were used. These combinations of molecules were presumably suitable for the reason of enhanced press formability. In this development, we confirmed that new composite coated steels are suitable for quality certification of digital TV panels. Table 1 shows diverse evaluation results for both conventional and new composite coated steels.

		Conventional	New composite
Coating Weight (mg/m²)	Bottom Layer	20±5 [Cr]	300±50 [Cr-free]
	Top Layer	1000±200	700±150
SST (hr)	Plane	72 ≤	96 ≤
	Drawing	4 8 ≤	72 ≤
Whiteness (L)		74	75
Anti-fingerprint (<i>\Delta E</i>)		0.25	0.15
Alkaline Resisitance (⊿E)		0.26	0.28
Chemical Resistance (∆E)	Iso-propanol	0.05	0.06
	Methylethylketon	0.22	0.20
	Methylene choride	0.13	0.14
	Toluene	0.08	0.10
Thermal Endurance (⊿E)		20.5	12.5
Abrasion Resistance (grade)		3	1
Friction Coefficient (a.u.)		0.120~0.130	0.115~0.120
Electrical Conductance $(m\Omega)$		≤ 0.045	≤ 0.045

Table 1. Summary of quality evaluation of conventional and new composite coated electro-galvanized steels

3.2.1 Corrosion Resistance. Corrosion resistance tests of conventional and new composite coated steel products are shown in Figure 4 and compared corrosion appearance after salt spray test. The composite coated products with coating weights of 300 and 700 mg/m² for the bottom and top layer, respectively, show excellent corrosion resistance in Ericksen drawn part from the salt spray test. This result is compared well with conventional anti-fingerprint steels with coating weights of 100 and 1000 mg/m² for the bottom and top layers, respectively. This suggests that the formation of organic-inorganic hybrid layers among zinc surface and double layers gives an excellent barrier and self-healing effect.

3.2.2 Electric Conductivity. In household appliances surface electric conductivity is important for electrical grounding of the instruments to shield electromagnetic interference. Therefore, a small coating weight is usually adopted to overcome the drawback of the

insulating property of the organic coating layer in the surface coated steels. The new composite coated steels exhibit surface electrical resistances less than 0.045 m Ω , which is acceptable for use in electronic appliances. These results suggest that strong binding between the organic-inorganic composite layer and zinc surface causes high surface electric conductance. New Cr-free composite coated steels that have a thin coating layer less than 1 μ m with coating weights of 1000±150 mg/m² assure excellent conductivity.





3.2.3 Thermal Endurance. Figure 4 shows the thermal endurance results of the composite coated steels compared with conventional anti-fingerprint steels. Usually, most organic resins cannot withstand at elevated heating temperature, more than 200 °C, because of thermal breaking or yellowing of the polymer. This new composite coated steels can withstand deterioration of the coating layer above the typical cure temperature up to 170 °C, even though slightly changed at 250°C. This result can be explained by the strong binding and thermal stability of both the inorganic bottom layer and urethane based composite top layer on the zinc surface.

3.2.4 Abrasion Resistance. Figure 4 shows the tribological tests for both steels and compared abrasion resistance. Even though there are similar coating weights between conventional and new composite coated steels, the latter exhibits better abrasion resistance properties as a 1st grade than 3rd grade when compared with the former. This finding can be explained by the strong binding and increased hardness of both the inorganic bottom layer and urethane based composite top layer on the zinc surface.

4. Conclusions

We have developed new Cr-free composite coated steel sheets to enhance abrasion resistance of conventional anti-fingerprint steels for application in home electronic appliances. The new composite coating solutions were prepared as an aqueous solution of both Cr-free inorganic solution instead of chromate for the bottom layer and polyurethane based composite solution instead of polyacrylate based composite solution for the top layer.

The composite coated steels were produced in the electro-galvanized line from the double layer coating with a roll coater. This composite steel exhibits excellent functionalities that have excellent corrosion and abrasion resistance, as well as surface conductivity within target coating weights. Experimental results suggest that strong binding among the inorganic bottom layer and organic-inorganic top layer on the zinc surface plays a key role in the excellent corrosion and abrasion performances. From the customer evaluation, new coated steels obtained quality certification for adaptation of back cover panels of LCD TVs, shown in Figure 5. The new composite coated anti-fingerprint steels are under mass commercial production in the POSCO EGL line. We pursue global sales by supplying a variety panels in the household electronic appliances such as digital TV.



Figure 5. Advanced abrasion property of new composite coated steels; customer evaluation of press formability (34" LED TV back cover, steel thickness 0.8mm).

5. References

[1] A. Gerd; H. Jost; H. Klaus; S. Christoph, K. Ulrich; K. Herbert; S. Peter; O. Manfred., "Chromium Compounds" *Ullmann's Encyclopedia of Industrial Chemistry*. Weinheim: Wiley-VCH, 2005.

[2] Official Journal of EU, L 37/19, 13, 2, 2003.

[3] Official Journal of EU, L 37/24, 13, 2, 2003.

[4] D.-H. Jo; M. Kwon; J.-S. Kim, "Chromate-free Hybrid Coating for Corrosion Protection of Electro-galvanized Steel Sheets" *Corrosion Science and Technology*, *Vol.11*, *No.(1)*, 15-19, 2012.
[5] M. Yamshita et al., "Growing Steel Sheets and Coated Steel Sheets", *The 167th and 168th Nishiyama Memorial Technical Symposium*, Tokyo, pp158-161, 1998.

[6] W.J. Van Ooij; A. Seth; T. Mugada; G. Pan and D.W. Scgaefer, "A Novel Self-Priming Coating for Corrosion Protection", *Proceedings of the 3rd International Engineering Congress*, Orlando, FL, August 2-4, 2004.

[7] P. Innocenzi and T. Kidchob; T. Yoko, "Hybrid Organic-Inorganic Sol-Gel Materials Based on Epoxy-Amine Systems", *J. of Sol-Gel Science and Technology*, 35, 225-235, 2005.

[8] D. F. Benjamin, "Roll Coating Flows and Multiple Roll Systems", *Ph.D. Thesis*, University of Minnesota, 1994.

[9] Japan Standards, JIS Z2371.